

**ATOMIC PRACTICE #2**  
**( $p^+$   $n^0$   $e^-$ , ATOMIC NUMBER, MASS NUMBER)**

**ATOMIC STRUCTURE**

For each atom, list the following, in order:

	ATOMIC #	MASS #	$p^+$	$n^0$	$e^-$
1)	Si-29				
2)	O-18				
3)	K-40				
4)	Li-7				
5)	Ni-58				
6)	Pm-147				
7)	H-3				
8)	Bi-209				
9)	Rn-222				
10)	Cu-65				

- 11) Draw at atom of Se-79, showing the particle totals for the nucleus and “shell” area.
  
  
  
  
  
  
- 12) How did the ancient Greeks view the atom controversy?
- 13) What is “atomic neutrality”?
- 14) Contrast mass number & atomic mass.